

Enantiomeric 1,2,3-Trithiane-Containing Alkaloids and Two New 1,3-Dithiane Alkaloids from New Zealand Ascidians

A. Norrie Pearce,[†] Russell C. Babcock,[‡]
Chris N. Battershill,[§] Gretchen Lambert,[⊥] and
Brent R. Copp^{*,†}

Department of Chemistry, and Leigh Laboratory,
School of Environmental and Marine Sciences,
University of Auckland, Private Bag 92019,
Auckland, New Zealand, National Institute of
Atmospheric Research, P.O. Box 14-901, Kilbirnie,
Wellington, New Zealand, and Department of
Biological Science, State University of Fullerton,
Fullerton, California 92834

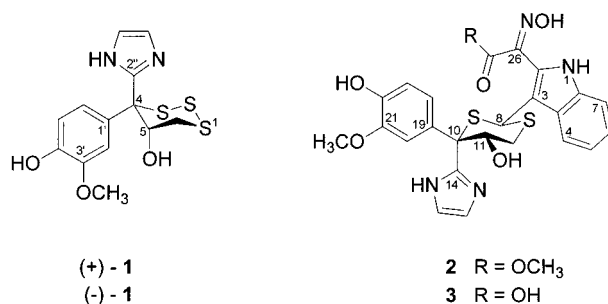
b.copp@auckland.ac.nz

Received July 31, 2001

Enantiomeric marine natural products are uncommon, and it is rare to find enantiomers with two or more chiral centers. Although such compounds are usually terpene-derived and typically reported from sponges,^{1–4} an ascidian-derived alkaloid enantiomeric pair, segolines B and C, have been recently reported.⁵

As part of our ongoing search for new bioactive compounds from New Zealand ascidians,⁶ we have found extracts of the delicate pink stalked ascidian, *Hypsistozoa fasmeriana* (Michaelsen, 1924) (order Aplousobranchia, family Holozoidae)⁷ to exhibit a wide range of biological activities. Bioassay-directed fractionation of extracts of specimens collected at Tutukaka, North Island, New Zealand, afforded the new (–) enantiomer of *trans*-5-hydroxy-4-(4'-hydroxy-3'-methoxyphenyl)-4-(2''-imidazolyl)-1,2,3-trithiane (**1**).⁸ A second collection of *H. fasmeriana* made at Leigh Harbor, Northland, afforded (–)-**1** and two novel dithiane alkaloids **2** and **3**.

All spectroscopic data observed for (–)-**1** were identical to the literature values originally reported for (+)-**1** with the exception of the chiroptical properties, which were equal in magnitude and opposite in sign at every wave-



(Does not imply absolute stereochemistry)

length measured (see the Experimental Section).⁹ To further investigate the enantiomeric nature of (–)-**1** and (+)-**1**, cyclodextrins (CD) were explored as chiral NMR shift reagents.^{10,11} NMR titrations of (+)-**1** with α -, β -, and γ -CD indicated that β -CD was the optimal host for **1** and that maximal upfield shifts of the CD protons H-3 and H-5 were observed when the ratio of β -CD to **1** reached 2:1.¹¹ Host–guest binding was confirmed by the observation of ROESY NMR (τ_{mix} 300 ms) correlations between H-3 of β -CD and H-4''/5'', H-5', and H-6' of **1**. These results combined with the observation of only a single set of resonances for each of β -CD and **1** in the mixture indicated that the host and guest were in rapid exchange on the NMR time scale. A racemic mixture of **1** was then prepared by combining both enantiomers of **1**, and as expected, the ¹H NMR spectrum of this mixture was indistinguishable from the spectra observed for both pure enantiomers. However, when β -CD was added to the racemic mixture, all resonances (with the exception of **1** H-5) doubled, consistent with the formation of diastereomeric host–guest complexes (see the Supporting Material).

A molecular formula of C₂₆H₂₄N₄O₆S₂ for **2** was established by HRFAB mass spectrometry. Several features of the ¹H NMR spectrum of **2** were similar to **1**, including the presence of 3-methoxy-4-hydroxyphenyl- and 2-substituted imidazole rings and a CH₂CH(OH) spin system. Connections between these three fragments were established by HSQC and HMBC NMR experiments allowing construction of the C-10 to C-25 portion of **2**. In addition to these fragments, NMR data were observed consistent with the presence of a 2,3-disubstituted indole ring (NH-1 to C-7), an isolated methine (C-8, δ_{H} 5.85, δ_{C} 42.8), and a carboxymethyl ester fragment (δ_{H} 4.02, δ_{C} 54.5 and 162.0, ν 1745 cm^{–1}). HMBC correlations observed for H-8 (δ 5.85) allowed connection of C-8 to indole C-3 (δ 97.2). Chemical shift considerations required C-8 (δ 42.8), C-10 (δ 59.2), and C-12 (δ 36.0) to be sulfur substituted, while long-range ¹H–¹³C correlations from H-8 to C-10 and C-12 established the presence of a 2,4,4,5-tetrasubstituted 1,3-dithiane ring. This left the carboxymethyl ester and a molecular fragment of CNOH unaccounted for, which must be present at the indole C-2 position. A weak ⁴J_{CH} HMBC NMR correlation from the methyl ester

* To whom correspondence should be addressed. Tel.: 64-9-373-7599 ext 8284. Fax: 64-9-373-7422.

[†] Department of Chemistry, University of Auckland.

[‡] Leigh Laboratory, SEMS, University of Auckland.

[§] National Institute of Water and Atmospheric Research. Current address: Australian Institute of Marine Science, PMB No 3, Townsville MC, Queensland 4810, Australia.

[⊥] California State University Fullerton. Current address: 12001 11th Ave NW, Seattle, WA 98177.

(1) Fontana, A.; Fakhr, I.; Mollo, E.; Cimino, G. *Tetrahedron: Asymmetry* **1999**, *10*, 3869–3872.

(2) Searle, P. A.; Jamal, N. M.; Lee, G. M.; Molinski, T. F. *Tetrahedron* **1994**, *50*, 3879–3888.

(3) Horton, P.; Inman, W. D.; Crews, P. *J. Nat. Prod.* **1990**, *53*, 143–151.

(4) Capon, R. J.; MacLeod, J. K. *Aust. J. Chem.* **1988**, *41*, 979–983.

(5) Viracaoundin, I.; Faure, R.; Gaydou, E. M.; Aknin, M. *Tetrahedron Lett.* **2001**, *42*, 2669–2671.

(6) Copp, B. R.; Wassvik, C. M.; Lambert, G.; Page, M. J. *J. Nat. Prod.* **2000**, *63*, 1168–1169.

(7) All ascidians used in this study were identified by one of us (G.L.).

(8) Copp, B. R.; Blunt, J. W.; Munro, M. H. G.; Pannell, L. K. *Tetrahedron Lett.* **1989**, *30*, 3703–3706. The systematic name of the (+) enantiomer was incorrectly reported in this paper as the *cis* configuration.

(9) (+)-**1** was reisolated from *Aplidium* sp. D, kindly supplied by Professor Munro of The University of Canterbury, to allow direct comparison of optical rotation data observed for both enantiomers of **1**.

(10) Li, S.; Purdy, W. C. *Chem. Rev.* **1992**, *92*, 1457–1470.

(11) Greatbanks, D.; Pickford, R. *Magn. Reson. Chem.* **1987**, *25*, 208–215.

Table 1. ^1H and ^{13}C NMR Data (CD_3OD) for Fasmerianamines **2** and **3**

atom	2		3	
	^1H (δ , mult, J)	^{13}C	^1H (δ , mult, J)	^{13}C
1	11.56 (s) ^a		11.53 (s) ^a	
2		131.3 (1, 8) ^{b,c}		133.1 (1, 8)
3		97.2 (1, 4, 8)		94.8 (1, 4, 8)
3a		126.1 (1, 5, 7)		126.2 (1, 5)
4	7.85 (dd, 7.1, 1.2)	120.3 (6)	7.76 (dd, 6.5, 1.9)	119.7 (6)
5	7.10 (ddd, 7.2, 7.2, 1.1)	121.1 (7)	7.08 (ddd, 7.2, 7.2, 1.4)	121.1 ^c
6	7.14 (ddd, 7.2, 7.2, 1.3)	123.3 (4)	7.10 (m)	122.7 (4)
7	7.38 (dd, 7.2, 1.0)	112.6 (5)	7.34 (d, 7.2)	112.5 (5)
7a		134.7 (1, 4, 6)		134.3 (1, 4, 6)
8	5.85 (s)	42.8 (12)	5.87 (s)	42.3 (12)
10		59.2 (8, 11, 12, 20, 24)		59.3 (11, 12, 20, 24)
11	5.07 (dd, 11.2, 3.3)	75.3 (12)	5.10 (dd, 11.1, 3.1)	75.3 (12)
12 α	2.98 (dd, 14.5, 3.3)	36.0 (8, 11)	2.95 (dd, 14.5, 3.1)	36.0 (11)
β	3.16 (dd, 14.5, 11.2)		3.16 (dd, 14.5, 11.1)	
14		147.9 (11, 16, 17)		148.0 (11, 16, 17)
16	7.69 (s)	121.4 (17)	7.69 (s)	121.4 (17)
17	7.69 (s)	121.4 (16)	7.69 (s)	121.4 (16)
19		131.4 (11, 23) ^c		131.3 (11, 23)
20	6.90 (d, 2.3)	111.7 (24)	6.94 (d, 2.3)	111.7 (24)
21		149.4 (20, 23, 25)		149.4 (20, 23, 25)
22		148.6 (20, 23, 24)		148.5 (20, 23, 24)
23	6.71 (d, 8.5)	116.4	6.69 (d, 8.4)	116.2
24	6.20 (dd, 8.5, 2.3)	121.0 (20)	6.16 (dd, 8.4, 2.3)	121.1 (20) ^c
25	3.86 (s)	56.6	3.85 (s)	56.8
26		156.0 (28)		155.3
27		162.0 (28)		161.2
28	4.02 (s)	54.5	13.10 (br s) ^a	

^a DMSO-*d*₆. Broad singlets were also observed at δ 12.19, 10.93, and 9.25 for **2** and at δ 10.93 and 9.25 for **3**. Due to their line widths the signals could not be assigned to specific protons in the molecules. ^b Numbers in parentheses are protons to which the carbon correlates in HMBC NMR experiments ($^1J_{\text{CH}} = 8.3$ Hz). ^c Assignments may be interchanged.

protons (δ 4.02) to an sp^2 -hybridized carbon at δ 156.0 established the presence of an unusual oximo acetate functional group. ROE correlations between H-11 (δ 5.07) and H-20 (δ 6.90), H-24 (δ 6.20) and H-12 α (δ 2.98), and H-8 (δ 5.85) and H-12 β (δ 3.16) secured the relative stereochemistry and the structure of **2** as (8 α ,10 β ,11 β)-methyl-2-[3-[11-hydroxy-10-(22-hydroxy-21-methoxyphenyl)-10-(18H-imidazol-14-yl)-9,13-dithian-8-yl]-1H-indol-2-yl]-2-oximoacetate.

Fasmerianamine B (**3**) was isolated as a 1:2 mixture with **2**. The NMR data observed for **3** were almost identical to those observed for **2**, with the greatest differences being an absence of the methyl singlet at δ 4.02 and the appearance of a broad exchangeable resonance at δ 13.10 suggesting the presence of a free carboxylic acid in place of the methyl ester. This was confirmed by the molecular formula ($\text{C}_{24}\text{H}_{22}\text{N}_4\text{O}_6\text{S}_2$) determined by HRFABMS and by the observation of strong IR absorptions at 3200 cm^{-1} (br) and 1685 cm^{-1} .

In our HPLC screening of New Zealand ascidians ($N = 130$), we have detected **1** in three different species. The (+)-enantiomer has been identified in *Aplidium* sp. D (order Aplousobranchia, family Polyclinidae),⁸ collected at Kaikoura, South Island, NZ, an unidentified *Aplidium* species from Stewart Island, NZ,¹² and *Distaplia stylifera* (order Aplousobranchia, family Holozoidae), collected at Cape Reinga, Northland, NZ. To date, however, the (–)-enantiomer of **1** has only been detected in *H. fasmeriana*. Given the close phylogeny of ascidians of the genera *Hypsistozoa* and *Distaplia* (both family Holozoidae), it is interesting to speculate on the biosynthetic divergence that is capable of producing enantiomers of alkaloids containing two stereogenic centers. One possible

biogenesis of **1** is via enantiomeric 5-oxo analogues, which then undergo reduction directed by the neighboring imidazole group at C-4.

Both enantiomers of **1** exhibited identical biological activities in a range of assays, including modest cytotoxicity (P388 IC₅₀ 21.6 μM) and antimicrobial (*Bacillus subtilis*, *Candida albicans*) properties (4 mm zone of inhibition at 120 $\mu\text{g}/\text{disk}$). Compound (+)-**1** was found to be inactive at the NCI (panel average values of GI₅₀ 32 μM , TGI 63 μM , LC₅₀ 100 μM), did not inhibit the cell cycle checkpoint enzyme cdc2/cyclin B kinase (at 100 μM), did not induce thiol-generated DNA strand scission,^{13,14} and was inactive (10% inhibition at 12.5 $\mu\text{g}/\text{mL}$) against *Mycobacterium tuberculosis* H₃₇Rv. Alkaloids **2** and **3** were inactive in all assays.

Experimental Section

Collection, Extraction, and Isolation Procedures. Specimens of *H. fasmeriana* were collected by scuba at Leigh Harbor and Tutukaka and kept frozen until use.⁷ Specimens of *D. stylifera* were collected by scuba at Cape Reinga, Northland, NZ, and kept frozen until use.⁷ Voucher specimens are retained at the University of Auckland, Department of Chemistry (*H. fasmeriana*, 99LH3-1), and at the NIWA Museum, Wellington (*D. stylifera*, 99MNP0266). *Aplidium* sp. D was collected at Kaikoura, South Island, NZ, and kept frozen until use. The freeze-dried (4.55 g) specimens of *H. fasmeriana* collected at Tutukaka were extracted with MeOH followed by CH_2Cl_2 . The crude extract was partitioned between water and CH_2Cl_2 and the aqueous fraction was subjected to C18 flash chromatography (0.1% aqueous TFA through to MeOH) with the compound of interest eluting at 25% MeOH. This compound was further purified by semipreparative C18 HPLC (4 mL/min, 0.05%

(13) Mitra, K.; Kim, W.; Daniels, J. S.; Gates, K. S. *J. Am. Chem. Soc.* **1997**, *119*, 11691–11692.

(14) Matsumoto, S. S.; Sidford, M. H.; Holden, J. A.; Barrows, L. R.; Copp, B. R. *Tetrahedron Lett.* **2000**, *41*, 1667–1670.

(12) Copp, B. R. PhD Thesis, Chapter 4, The University of Canterbury, New Zealand, 1989.

aqueous TFA+MeOH (6:4)) yielding (–)-**1** as the TFA salt (8 mg, 0.18% dry wt). Freeze-dried specimens of *Aplidium* sp. D (17.79 g) were worked up in the same manner yielding 45.3 mg (0.25% dry wt) of (+)-**1** as the TFA salt. Specimens of *H. fasmeriana* collected from Leigh Harbor were freeze-dried (3.50 g) and extracted with MeOH and CH₂Cl₂. The crude extract was partitioned between water and CH₂Cl₂, and the aqueous fraction was subjected to repeated C18 flash chromatography (0.1% aqueous TFA through to MeOH) eluting fasmerianamine A (**2**) (0.97 mg, 0.03% dry wt) and fasmerianamine B (**3**) (2.4 mg, 0.07% dry wt) as a 2:1 mixture of **2** and **3**.

(–)-**1**: [α]_D²⁰ (λ) –250 (365 nm), –33 (546), –26 (578), –26 (Na D line) (c = 0.1, MeOH); MS (DEI) m/z 342 (1), 310 (4), 246 (8); HREIMS calcd for C₁₃H₁₄N₂O₃S₃ 342.0167, found 342.0146; IR (film) ν_{\max} 3400, 1673, 1597, 1522, 1426, 1201, 1134 cm^{–1}; UV (MeOH + TFA) λ_{\max} (log ϵ) 285 (3.34) nm; ¹H NMR (CD₃OD, 400 MHz) δ 7.54 (2H, s, H-4'', H-5''), 6.97 (1H, d, J = 2.3 Hz, H-2'), 6.79 (1H, d, J = 8.4 Hz, H-5'), 6.62 (1H, d, J = 2.3, 8.4 Hz, H-6'), 5.26 (1H, t, J = 7.2 Hz, H-5), 3.83 (3H, s, 3H-7'), 3.70 (1H, dd, J = 7.2, 11.0 Hz, H-6 β), 2.94 (1H, dd, J = 7.2, 11.0 Hz, H-6 α); ¹³C NMR (CD₃OD, 100 MHz) δ 149.6 (C-3'), 149.1 (C-4'), 147.8 (C-2''), 127.4 (C-1'), 121.7 (C-6'), 120.9 (C-4'', C-5''), 116.8 (C-5'), 111.7 (C-2'), 84.2 (C-5), 67.9 (C-4), 56.6 (C-7'), 39.8 (C-6).

(+)-**1**: [α]_D²⁰ (λ) +260 (365 nm), +34 (546), +28 (578), +26 (Na D line) (c = 0.1, MeOH); FTIR, UV, MS, ¹H NMR, and ¹³C NMR data were identical to those of (–)-**1**.

Fasmerianamine A (2): brown gum; [α]_D²⁰ = –17.5 (c = 0.4, MeOH); MS (FAB) m/z 541 (MH⁺, 3); HRFABMS calcd for C₂₆H₂₅N₄O₆S₂ 541.1216, found 541.1224; IR ν_{\max} (film) 3305, 1745 (sh), 1677, 1432, 1202, 1136 cm^{–1}; UV (MeOH) λ_{\max} (log ϵ) 206 (4.13), 283 (3.35), 311 (3.18) nm; UV (MeOH + TFA) λ_{\max} (log ϵ) 207 (4.20), 285 (3.36), 311 (3.18) nm; UV (MeOH + KOH) λ_{\max} (log ϵ) 208 (4.42), 253 (3.65), 342 (3.25) nm; ¹H NMR (CD₃OD) see Table 1; ¹³C NMR (CD₃OD) see Table 1.

Fasmerianamine B (3): brown gum; MS (FAB) m/z 527 (MH⁺, 0.5); HRFABMS calcd for C₂₄H₂₃N₄O₆S₂ 527.1055, found

527.1059; IR ν_{\max} (film) 3188 (br), 1685 cm^{–1}; ¹H NMR (CD₃OD) see Table 1; ¹³C NMR (CD₃OD) see Table 1.

Acknowledgment. We gratefully acknowledge funding from the Auckland Medical Research Foundation and the University of Auckland Research Committee. We wish to thank Professor Murray Munro and Mrs. Gill Ellis of the University of Canterbury for supplying freeze-dried specimens of *Aplidium* sp. D, Mrs. Gill Ellis for the P388, cytotoxicity and antimicrobial assays, and Dr. V. Narayanan (NCI, Bethesda) for in vitro human antitumor assays. We also thank Dr. Sandra Matsumoto and Professor Louis Barrows of the University of Utah for the DNA cleaving studies, Dr. Laurent Meijer (CNRS, France) for the cdc2/cyclin B kinase assay result, the Tuberculosis Antimicrobial Acquisition and Coordinating Facility (TAACF) for antimycobacterial assays, and the Coral Reef Foundation (under contract from the NCI, Bethesda; supported by NIWA (NZ FRST Contract No. C01809)) for collection of *Distaplia stylifera* specimens. Professor John Mann (Queen's University Belfast) is thanked for helpful discussions regarding the putative biogenesis of **1**.

Supporting Information Available: ¹H and ¹³C NMR spectra of (–)-**1**, ¹H NMR spectrum of *rac*-**1** + β -CD, ¹H spectrum of **2**, and ¹H and ¹³C NMR spectra of **3** as a 1:2 mixture with **2**. This material is available free of charge via the Internet at <http://pubs.acs.org>.

JO010769+